

according to 91/155/EEC including Directive 93/112/EC and Directive 2001/58/EC

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TUNGSTEN ELECTRODE LWT

1 Identification of Substance, Preparation and Company

Identification of product: TUNGSTEN ELECTRODE

Use of product: Non-melting electrode in TIG welding process; electrodes for light

engineering; electrodes for plasma smelting; plasma cutting, plasma spraying (thermal spraying); emission cathodes for electronic pipes.

Emergency phone number: Umweltbundesamt (Federal Environmental Agency), room 2.4 (GSA)

Bismarckplatz 1; 14193 Berlin, Germany

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2 Composition / Data on Components

Substance	Formula	CAS No.	EINECS No.	Molecular weight	Classific. accord. to DI. 67/548/EEC
Tungsten	W	7440-33-7	231-143-9	183.85 g/mol	none
Lanthanum oxide	La₂O₃	1312-81-8	215-200-5	325.82 g/mol	none
Cerium oxide	Ce ₂ O ₃	1345-13-7	215-718-1	328.24 g/mol	none
Zirconium oxide	ZrO ₂	1314-23-4	215-227-2	123.22 g/mol	none
Y ttrium oxide	Y ₂ O ₃	1314-36-9	215-233-5	225.82 g/mol	none

3 Potential Dangers

This product does not feature hazardous properties for the purpose of the EU directive 67/548/EEC (Dangerous Substances) and 99/45/EC (Dangerous Preparations regulations) and of the Act on the Protection against Hazardous Substances (Chemicals Act) of J une 2002.

Danger warning for humans:

Any precautions usually applied for the handling of chemicals must be

complied with. Avoid the development of dust and smoke. Do not inhale

dust and smoke.

Danger warning for environment: Waste has to be disposed of in a safe and secure manner. The interna-

tional regulations apply.

Classification system: not listed

4 First Aid Measures

After contact with the eyes: Rinse eyes for several minutes under running water, with the eye lid

open. Consult a doctor if condition persists.

After contact with skin: In general, the product is not skin-irritating. Remove dust thoroughly by

washing with soap.

After inhaling: Remove the affected person from the danger area and consult a doctor.

After ingesting: Thoroughly rinse mouth and consult doctor.



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FOR ALL CASES - CONSULT A MEDICAL SPECIALIST IF YOU ARE EXPERIENCING STRONG SYMPTOMS.

5 Fire Fighting Measures

General information The metal in compact form is not flammable.

Appropriate extinguishing agents: Extinguishing powder class D, water jet, sand

NOT appropriate extinguishing agents: Water, ABC powder, Halon, CO₂

Special risks: Dangerous disintegration products, see chapter 10

Special protection equipment: During fire fighting, self-contained breathing apparatus is recommended which complies with the regulations for use under positive pressure.

6 Measures for Accidental Release

Precautions for persons: For release of dust or smoke – extracting devices and breathing protec-

tion with particle filter P2 or P3, recommended P3, colour code: white¹.

Measures for protection of environment: Procure to avoid release into environment. Waste, dust filters and

recipients are to be disposed of in a safe and secure way, according to the respective national regulations in force. Grey washing or grinding

water is to be captured and disposed of.

Method for cleaning/retaining: mechanical sweeping

7 Handling and Storage

Handling Avoid the incorporation of dust during processing by use of appropriate

extraction devices and breathing protection with particle filter P2 or P3;

recommended P3, colour code: white. Avoid the formation of dust.

Storage store in a dry place

Intended use

This product is intended for use as a non-melting electrode for the TIG

welding process. Dust and smoke emerging during this process must

welding process. Dust and smoke emerging during this process must be extracted by means of appropriate devices featuring filters or gas

scrubbers.

8 Restriction of Exposure and Personal Protection Equipment

Exposure limits: Germany

Dust exposure TRGS 900²

Identity of substance		limit	notes
Identification	EU no. CAS no.	mg/m³	
Tungsten	231-143-9 7440-33-7	5 E	DK, 25

Austria MAK	Daily medium 5 mg/m³ Short term value 10 mg/m³		
Denmark OEL	TWA: 5 mg (week)/m³		J an 1999 ³
Netherlands MAC	TWA (8h) 5 mg/m ³		2002 ⁴
Poland OEL	MAC (TWA): 5 mg (week)/m³		J an 1999 ³
Russia OEL		STEL: 2 mg/m³	J an 1999³



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Norway OEL	TWA: 5 mg (week)/m³		J an 1999³
Sweden OEL	NGV: 5 mg (week)/m³		J an 1999 ³
UK OEL	TWA 5 mg (week)/m³	STEL:10 mg (week)/m³	Sep 2000 ³
USA, NIOSH, REL	TWA 5 mg (week)/m³	STEL: 10mg/m³	DHHS,1992 ³
USA, MSHA	TWA 5 mg (week)/m³		DTLVS,1972 ³
USA, ACGIH, REL	TWA 5 mg (week)/m³	STEL:10 mg/m³	RTK#1959 ⁵

Restriction and supervision of exposure

Exposure: General Change of contaminated clothing; Washing of hands after handling,

possibly also showering

Keep away from food, beverages and animal feed.

Exposure: At workplace

PSA breathing protection Extraction device, mask with particle filter (protection class P2) recom-

mended for dust/aerosols. Protection class and type of mask must be adjusted according to the actual dust exposure, especially during clean-

ing and maintenance work⁶.

PSA hand protection UV protection welding gloves, general protection and hygiene meas-

Eye protection Goggles, face protection shield recommended.

Body protection Ban on food, drinks and smoking in the workplace in connection with

work hygienic measures, e.g. washing hands.

Exposure: Environment Remaining items, residues have to be disposed of according to the

respective national regulations in force.

Physical and Chemical Characteristics

General information

Form: solid colour: metallic grey odour: inodorous melting point: 3,680 K boiling point: 5,828 K ignition temperature: not existent oxidising characteristics not applicable 0 hPa

vapour pressure at 20℃ (mm Hg):

Density at 20℃:

18.5 g/cm³

Insoluble in water; insoluble in fat; very acid-resistant; slowly soluble in Solubility:

HNO₃ + HF; soluble in alkaline oxidation melts

 $18.20 \pm 0.2 \text{ m/}\Omega \text{mm}^2$ Electrical conductivity



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Stability and Reactivity

Substances to be avoided:

Stability: Product is stable under normal conditions. No disintegration when used

Conditions to be avoided:

Presence of oxygen and high temperatures (>600 $^{\circ}$ C) cause oxidation, from 977° sublimation (tungsten trioxide WO 3, CAS 1314-35-8).

> Contact with strong acids and/or bases; or with halogens (fluorine, chlorine, bromine, iodine and their compounds); or with oxidation agents (e.g. perchlorate, peroxide, permanganate, chlorate, nitrates, nitrites, chromates); or with alkali/earth alkali metals (e.g. lithium, sodium, potassium; magnesium, calcium) can cause strong reactions (danger of strong exothermal reactions, danger of formation of flammable gases, danger of formation of insalubrious / poisonous

substances / gases), and must be avoided.

Dangerous disintegration products: Emerge through oxidation oxides of the product, which can evaporate

(tungsten trioxide WO₃, CAS 1314-35-8) or be released.

Toxicological Information

Acute toxicity: This product does not feature or just slight acute oral, dermal toxicity or toxicity through inhalation.

> LD₅₀ oral, rat: >2000 mg/kg⁷ LD₅₀ dermal, rat: >2000 mg/kg⁷ LC₅₀ inhalative, rat: >5.4 mg/l, 4h eposition⁷

La₂O₃ Clear epidemiologically secured data concerning lanthanoides and particularly La2O3 caused vocational illnesses are not present so far. LD50 oral, rat: 5 g/kg8.

 Ce_2O_3 Substance-specific data for acute effect particularly of Ce_2O_3 are missing widely. With oral application all examined inorganic cerium compounds, also the soluble Cer(III) salts, have shown a small acute toxicity^s

Y₂O₃ With an attempt at rodents signs of an acutely toxic effect (accelerated respiration) were observed after a 4h-Inhalation of concentrations off approx. 32 mg/m3. LD50 oral, rat: 5 g/kg

Quantitative data to the toxicity are not present. After inhalation of dust: Irritations of the respiratory. Dangerous characteristics are improbable 11 .

Further information from literature, see 3

Findings after intratracheal application of 50 mg tungsten dust/week during 3 weeks on guinea pigs lead to the conclusion that the substance is relatively inert. Still, an effect on the lung tissue could be proved (interstitial cellular proliferation), which must not be neglected. W dust, which was given to very young rats with their food over 70 days in concentrations of 2; 5 or 10%, caused a 15% reduction of the development of the body weight in the female, but not in the male animals¹².

Due to its extent, the irritation caused by the product does not need to be classified $^{\!13}\!\!$.

of the eye: Due to its extent, the irritation caused by the product does not need to

be classified¹³

Sensitisation: No sensitising effects known⁷.

of skin:

Chronic toxicity:

Primary irritation



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12 **Ecological Data**

Amphibians: LC $_{50}$:2.9 mg/l (toad, gastrophryne carolinensis, 7d) 14 fish: LC $_{50}$:15.6 mg/L (rainbow trout, oncorhynchus mykiss, 28d) 14 Ecological toxicity:

Biodegradability: not applicable.

Tungsten compounds can be found in soils and water as wolframat Mobility:

> (e.g. WO_4^{2-}) and other polyanions. There are no reports on organic tungsten compounds. The extraction coefficient for tungsten rises under the condition of decreasing ph-values (ph=5:100-50,000; ph=6.5:10-6,000; ph=8-9:5-90). These values prove slow or zero mobility of tungsten compounds in soils and water. In nature, tungsten compounds can be found as ions or insoluble solid matter. Therefore the volatilisation of surfaces of soils and water represents a less important environmental

Persistence and biodegradability:

Abiotic degradability:

Biodegradability: Not applicable.

> Tungsten features types of various oxidation values (0, 2+, 3+, 4+, 5+, 6+). The most stable type is 6+, the other types are quite unstable. As ion, tungsten exists in combination with one or more elements, e.g. with

oxygen. In water, tungsten compounds can be found as tungstate (e.g. WO_4^{2-}) and other polyanions. There are no reports on organic tungsten compounds. Bibasic tungsten only exists as a halogen compound. Tungsten strongly tends to form complexes (e.g. formation

of heteropoly acids with oxides of phosphor, arsine, vanadium, silicon, and more). Tungsten forms a series of oxohalogenides (e.g. WOCI₄).

Bioaccumulation potential: No data available

Other information: Water hazard class: not hazardous to water (German Water Hazard

Classes accord. to VwVwS (German Administrative Regulation for Sub-

stances Hazardous to Water) from 17th May, 1999)

Notes on Disposal

Waste disposal according to international, national and regional regulations. Contact your local office responsible for this.

This material and its container must be disposed of in a safe way. For

information on recycling/reutilisation, please contact the manufac-

turer/distributor.

Recommendation: Obey the national regulations for the disposal. Unclean packaging: Can be handled as non-hazardous waste.

Information on Transport

EU regulations Transport regulations do not apply to these products – no hazardous

material.

Specifications

Specification acc. to EEC directive: 215-225-1, 215-200-5, 215-718-1, 215-227-2, 215-233-5



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Directive 67/548/EEC idgF (Dangerous Substances) EU regulations:

Directive 99/45/EC idgF (Dangerous Preparations regulations)

German regulations: Technical Regulation Air: TRGS 900 Other countries: National provisions must be obeyed.

Other Information

This safety data sheet has been prepared for the described product and must only be used for the described product. The data refers to the current state of research and knowledge. It is meant to describe the product specified in this safety data sheet with regards to the required safety precautions. The data provided does not guarantee any characteristics of the described product. If this product is used as a component of another product or if it is modified by processing, the information in this safety data sheet may not be applicable. The conditions and methods of handling, storage, usage and disposal are not within our control. Due to these and other reasons we do not take responsibility and refuse any liability for reasons, the cause of which can be seen in handling, storage, usage or disposal of the product. The user is responsible for forwarding the information in this data sheet to the employee in an appropriate manner.

Sources

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